All Chemistry Formulas

Class IX

- Molecular mass
- ➤ Mole Concept
- ➤ Law of chemical combination
- Valency
- Concentration of solution

Class X

- pH valuepOH value
- Mole concept

Explanations

Class XI

- Atomic mass
- Average atomic mass
- > Atomic mass unit
- Molarity
- Molality
- Normality
- ➤ Ideal gas equation
 (Boyle's law ,Charles law ,Gay-Lussa's law, Avogadro's law)
- > Real gas equation

- > Daltons law of partial pressure
- ➤ VanderWaals equation
- First law of thermodynamics
- ➤ Work
- ➤ Heat capacity
- Entropy
- > Equilibrium constant
- ➤ Kinetic energy, potential energy, total energy
- ➢ Bohr's Radius

Class XII Explanations

- ➤ Henry's law
- ➤ Relative lowering of vapour pressure
- > Depression in freezing point
- > Elevation of boiling point
- Osmatic pressure
- Standard emf
- Nernst Equation
- ➤ Gibs free energy
- ➤ Kolhrauch's law
- Combined gas equation
- ➤ Integrated rate equation
- ➤ Differential rate & order
- ➤ Half life
- > Arrhenius equation

- > Freundlich adsorption Isotherm
- ➤ Langmuir Adsorption Isotherm

Net Explanations