**Chemical formula:** 

 $NH_3$ 

6) Formula of Metane: the elements present in this compound carbon and hydrogen.

Symbol:



Valency:

7} Formula of carbon dioxide: the elements present in this compound carbon and oxygen.

Symbol:



Valency:

Dividing by common factor: 2 1

> Formula:  $CO_2$

2} chemical formula of some simple ionic compound:

Formula of sodium chloride:

Symbol:



Charge:

Valency:

So, Na1+ and Cl1- charge is negligible or omitted. By the rules.

Chemical formula: NaCl

Formula of magnesium chloride:

Symbol:



1

Charge:

2

Valency:

Chemical formula: MgCl<sub>2</sub>

• Formula of Aluminium oxide: Al<sub>2</sub>O<sub>3</sub>

Valency: 3 2

Formula: Al<sub>2</sub>O<sub>3</sub>

3) Chemical formulae of compound containing polyatomic ion:

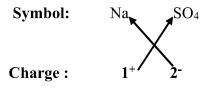
• Formula of potassium nitrates:

Formula: KNO<sub>3</sub>

Valency: 1 1

Chemical formula: KNO<sub>3</sub>

• Formula sodium sulphate: Na<sub>2</sub> SO<sub>4</sub>



Valency: 1 2

Chemical formula: Na<sub>2</sub> SO<sub>4</sub>

• Formula of calcium nitrate: Ca(NO<sub>3</sub>)<sub>2</sub>

Symbol: Ca  $NO_3$   $Charge: 2^+$   $1^-$ 

Valency: 2 1

Chemical formula: Ca(NO<sub>3</sub>)<sub>2</sub>

• Formula of Aluminium Sulphate: Al<sub>2</sub> (SO<sub>4</sub>)<sub>3</sub>

Symbol: Al SO.
Charge: 3+ 2-

Valency: 3 2

Chemical formula: Al<sub>2</sub> (SO<sub>4</sub>)<sub>3</sub>

**Key word: Chemical formula** 

**Question answer:** 

1) What is chemical formula?

**Answer:** Chemical formula is showing all constituents of chemical compounds by chemical symbol in a fixed proportion or according to the valency of elements or atom

2) Write the rules of writing chemical formula?

Answer: following are the rules of writing a chemical formula

• In case simple molecular compounds that is those which are made up of only two elements and hence also called binary compounds the symbol of